

**MARK SCHEME for the October/November 2012 series**

**0652 PHYSICAL SCIENCE**

**0652/51**

Paper 5 (Practical Test), maximum raw mark 30

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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- 1 (a) (i) entry for  $d$  for 50 g (must be < 60) ; [1]
- (ii)/(iii) remainder of entries for  $d$  (60, 70, 80, 90 g) ;  
all readings to nearest cm or all to nearest 0.1 cm (*consistency*) ;  
 $d$  values decrease for increasing  $m$  ; [3]
- (b) (i) all three  $1/m$  values:  
0.017/0.0167 (not 0.016)  
0.014/0.0143 (not 0.0142)  
0.013/0.0125 (not 0.012) ; [1]
- (ii) vertical axis linearly numbered **AND** labelled ;  
4 points plotted correctly within  $\frac{1}{2}$  square ;  
best straight line ;  
(*no graph marks for plotting wrong column from table but allow gradient to be calculated from a straight line*) [3]
- (iii) working shown either in space **or** on graph as coordinates, triangle **or**  $\Delta x$  and  $\Delta y$  **AND** change in  $d$  must be at least 10 (or 4 cm of paper vertically) ;  
gradient value from a correct working method ;  
(*no gradient marks from a graph with a curve or point to point lines*) [2]
- (iv) value using mass of rule = 300 – (gradient from (b) (iii)/10) ; [1]
- (c) (i) all mass  $\times$  distance values calculated and entered in table ;  
(*allow if only four masses in table*) [1]
- (ii) average mass  $\times$  distance value ; [1]
- (iii) value for mass of rule ; [1]
- (d) advantage of plotting shows anomalous results clearly ; [1]

[Total: 15]

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- 2 (a) (i) first value entered in column 2 of table and  $< 10$  ; [1]
- (ii) two more readings in column 2 ;  
all readings to 1 decimal point ;  
2 of the readings within  $0.4 \text{ cm}^3$  ; [3]
- (iii) column 3 completed (10 – column 2) ; [1]
- (iv) average calculation for  $V_{av}$  ; [1]
- (v) correct values used ( $V_{av} = \text{(a) (iv)}$ ,  $c_a = 0.013$  and  $V_a = 10$ ) ;  
correct rearranging  $c_s = 2 \times c_a \times V_a/V_{av}$  ;  
correct  $c_s$  calculated value to 2 (or more) significant figures ;  
(correct value only scores all 3 marks)  
(calculation mark may be awarded following wrong substitution and/or wrong rearrangement providing all terms included) [3]
- (b) (i) colour = red/orange **AND** pH = 1 – 4 ; [1]
- (ii) colour = yellow (or orange if (b) (i) is red) **AND** pH > (b) (i) pH and < 7 ; [1]
- (iii) colour = yellow/green **AND** pH = 6 – 7 but not < (b) (ii) pH ; [1]
- (c) colour = purple **AND** pH = 10 – 14 ; [1]
- (d) (calcium hydroxide because)  
2 spatula loads calcium carbonate and still not neutral (reference to (b)) ;  
1 spatula load calcium hydroxide produced greater increase in pH (ref to (c)) ;  
**OR**  
'1 spatula load calcium hydroxide produced greater increase in pH than 1 spatula load calcium carbonate' (scores 2 marks) ; [max 2]

**[Total: 15]**